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Overview: Diverse
Forms, Common
Challenges. Animals
inhabit almost every
part of the biosphere.
Despite their great

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diversity, all animals must solve a common set of problems. All animals must obtain oxygen, nourish themselves, excrete wastes, and move.

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Chapter 40: Basic Principles of Animal Form and Function 1. Explain how amoebas, hydras, tapeworms, and whales exchange

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materials with their environment.

Exchange occurs as substances dissolved in an aqueous solution move across the plasma membrane of each cell. The

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Concept 40.1 → Animal form and function are correlated at all levels of organization

Evolution of animal

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size and shape The range of animal forms is limited by the physical laws that govern strength, diffusion, movement, and heat exchange.

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of an organism. physiology. the processes and functions of an organism. interstitial fluid. the fluid filling spaces between cells in most animals. tissues. an integrated group of cells with a common structure, function, or both ...

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Form and Function

Concept 40.1 Animal

Form and function are correlated at all levels of organization.

Physical laws, especially physics of exchanging materials with environment, limits range of animal forms

Exchange in Simple Organisms

Single celled organisms

- o Exchange material by diffusion
- o Entire

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Function Answer

Key

membrane is exposed
to medium Multicellular
organisms with a sac
body plan o Body wall
is only 2 cell layers
thick o Exchange is by
diffusion Fig 40.3 ...

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How do animal structures well suited to specific functions come about? A) Natural selection favors the most functional structures for a particular environment.

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approaches to conflict resolution: (1) conflict resolution is a cooperative endeavor, (2) the solutions sought are integrative ones, (3) the foundation is an understanding of all parties' interests, and (4) both the process and its outcome are nonviolent We will

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electrostatic attraction energy between ions of opposite charge is directly proportional to the charge on each ion (Q_1 and Q_2 in Equation 9.1). Thus, more energy is released as the charge on the ions increases (assuming the internuclear distance does not increase substantially).

1.12: Basic Concepts of Chemical Bonding

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