

## Basic Atomic Structure Worksheet Answer Key

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### Basic Atomic Structure Worksheet Answer

Basic Atomic Structure Worksheet ANSWERS. Basic Atomic Structure Worksheet ANSWERS 1 a) protons b) neutrons c) electrons a) Positive b) Neutral c) negative 2 atomic number (or identity); charge 3 protons; electrons (in a neutral charge atom only!); same 4 average atomic weight; mass 5 mass number; nucleus 6 neutrons; protons (or atomic number); mass number 7 Lithium = Li = 3 Iron = Fe=26 Oxygen = O = 8 Krypton = Kr = 36 8 (A neutral atom by definition has NO charge!

### Basic Atomic Structure Worksheet ANSWERS

Basic Atomic Structure Worksheet H and the 1. 2. 4. 5. 6. 8. 9. The 3 particles of the atom are: protons (p) neutrons (n) electrons (e) Their respective charges are: The number of protons in one atom of an element determines the atom's number Of electrons determines the of the element in one atom of an element. It also in a neutral atom of that element. The atomic number

### 090412 Atomic Structure Worksheet 1

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2. The number of protons in one atom of an element determines the atom's \_\_\_\_\_, and the number of electrons determines the \_\_\_\_\_ of the element.  
3.

### Basic Atomic Structure Worksheet

Write two algebraic equations relating the number of protons and neutrons in an atom to that atom's atomic mass and atomic number: Where,  $P$  = Number of protons.  $N$  = Number of neutrons.  $A$  = Atomic mass.  $Z$  = Atomic number

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### Atomic Structure Worksheet Answer Sheet

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### Neshaminy School District / Overview

Atomic Structure Worksheet. Define mass number: The mass number is an integer (whole number) equal to the sum (addition) of the number of protons and neutrons in an atomic nucleus. Define atomic number: The number of protons in the nucleus of an atom. The three subatomic particles of the atom are: Protons, electrons, neutrons

### Basic Atomic Structure Worksheet - Science - Home

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Atomic Structure MODULE - 2 Notes Atomic Structure and Chemical Bonding hemistry has been defined as the study of matter in terms of its structure, composition and the properties. As you are aware, matter is made up of atoms, and therefore an understanding of the structure of atom is very important. You have studied in your earlier

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