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Nuclear Chemistry Notes. I. Representation of an atom or nuclide: Radium-228 OR 228 (mass #) name of element followed by mass # Ra (symbol) ... So now we know the answer will be:  $4\text{mg} \times \frac{1}{2} \times \frac{1}{2} \times \frac{1}{2} = 0.25\text{mg}$ . Practice: 1. The half-life of polonium-210 is 138.4 days. How many mg of polonium-210 remain after 415.2 days if you start with 2mg ...

### **Nuclear Chemistry Notes and Worksheet**

1) Transition metals. Iron + bromine iron (III) bromide.  $2\text{Fe} (s) + 3\text{Br}_2 (l) \rightarrow 2\text{FeBr}_3 (s)$  Iron + chlorine iron (III) chloride.  $2\text{Fe} (s) + 3\text{Cl}_2 (g) \rightarrow 2\text{FeCl}_3 (s)$  All of the compounds with metals are ionic salts which form a giant structure. They are called metal halides because they are formed from a metal and a halogen.

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